

The Preparation and X-Ray Structure of Naphthalenedithiadiphosphetanedisulphide

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Reaction of P_4S_{10} with α -bromonaphthalene at 240 °C gave the title compound, (1), in 25% yield, the first example of a six-membered (C_3P_2S) organo-P-S ring, the X-ray structure of which reveals a novel folded P_2S_2 ring.

There is currently interest in the preparation of organic sulphur-nitrogen heterocycles.¹ However there are very few examples of organophosphorus-sulphur ring compounds. Phosphorus sulphides such as P_4S_{10} and $RP(S)_2P(S)R$ ($R = MeOC_6H_5 =$ Lawessons Reagent) are important in organic synthesis as thionating agents.² Apart from being of interest in their own right, new carbon containing phosphorus-sulphur heterocycles may offer the opportunity of variable selectivity in synthesis. Here we report the preparation of the first example of a six-membered organothiophosphate heterocycle. To our knowledge there are no known examples of this ring system although there has been a brief report on the preparation of $C_6H_4(PMe_3)_2S$ which is believed to have a structure containing a C_2P_2S ring.³

Reaction of P_4S_3 with S_8 in α -bromonaphthalene at 240 °C is reported⁴ to give P_4S_9 in moderate yield. However the reaction is very sensitive to the conditions used. We have found that in order to maximise the yield of P_4S_9 a slightly lower temperature (*ca.* 220 °C) is preferable. If the temperature is raised to 240 °C, and above, compound (1) is obtained after cooling and addition of diethyl ether. Alternatively, (1) may be prepared in 25% yield by reaction of P_4S_{10} with bromonaphthalene. Reaction of P_4S_{10} with naphthalene also gives (1) but in lower yield together with at least five other compounds which we have not yet identified.

The X-ray structure of (1) (Figure 1)[†] shows the naphtho(PS)₂ unit to be planar (all of the atoms lying within a crystallographic mirror plane) and symmetrically bridged by two sulphur atoms. The molecule has, within statistical significance, C_{2v} symmetry. The geometry of the naphthalene ring and the C-P and P=S distances are as expected. However, the geometry of the P_2S_2 ring is noteworthy. The ring is non-planar with a dihedral angle of 135° about the $S \cdots S'$ axis. The two S-P-S angles [91.6(1), 91.8(2)°] and the P-S-P angles [80.0(1)°] are noticeably reduced relative to those in $RP(S)_2P(S)R$ systems [$R = Me,$ ⁵ Ph,⁶ 2,4,6-Me₃C₆H₂⁷ S-P-S 94.5(2), 93.1(1), and 93.8(1), P-S-P 85.5(2), 86.9(1), and 86.2(1)°, respectively]. The transannular $S(19) \cdots S(19')$ distance is 3.05 Å.‡ The molecules pack parallel to each other (interplanar separation between naphtho rings 3.56 Å) though with a minimum of overlap between adjacent naphtho rings.§

In view of the usefulness of Lawessons Reagent in organic synthesis we have investigated the ability of (1) to behave as a thionating agent. Both triphenylphosphine oxide and benzophenone are readily converted to the corresponding sulphide. Reaction of (1) with triphenylphosphine proceeds at room temperature to give $SPPH_3$ and an as yet unidentified ring system.

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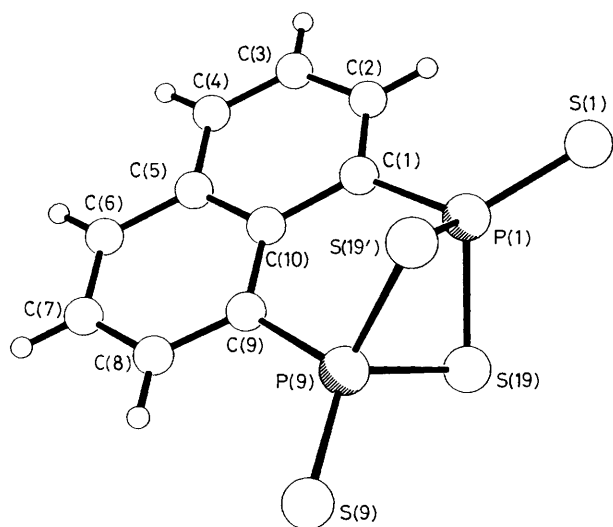


Figure 1. The X-ray crystal structure of (1). Important bond distances and angles: C(1)-P(1) 1.805(6), C(9)-P(9) 1.809(6), P(1)-S(1) 1.913(2), P(9)-S(9) 1.912(2), P(1)-S(19) 2.126(2), P(9)-S(19) 2.122(1), S(19) \cdots S(19') 3.05 Å; C(1)-P(1)-S(1) 117.5(2), C(9)-P(9)-S(9) 116.9(2), C(1)-P(1)-S(19) 103.7(1), C(9)-P(9)-S(19) 103.6(1), S(1)-P(1)-S(19) 118.2(1), S(9)-P(9)-S(19) 118.6(1), S(19)-P(1)-S(19') 91.6(1), S(19)-P(9)-S(19') 91.8(1), P(1)-S(19)-P(9) 80.0(1)°.

[†] Crystal data for (1): $C_{10}H_6P_2S_4$, $M = 316.4$, orthorhombic, $a = 7.116(1)$, $b = 9.209(1)$, $c = 9.494(3)$ Å, $U = 1244$ Å³, space group $Pm\bar{c}n$, $Z = 4$, $D_c = 1.69$ g cm⁻³, $\mu(Cu-K\alpha) = 92$ cm⁻¹. Data were collected on a Nicolet R3m diffractometer using graphite-monochromated Cu-K α radiation, ω -scans. The structure was solved by direct methods, corrected for absorption, and refined anisotropically to give $R = 0.048$, $R_w = 0.051$ for 834 independent observed reflections [$|F_o| > 3\sigma(|F_o|)$, $\theta \leq 58^\circ$]. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

[‡] The shortest intermolecular $S \cdots S$ contacts are between S(9) and S(19), 3.38 Å, and S(1) and S(9), 3.48 Å.

[§] In adjacent molecules H(2') lies approximately over the centre of the C(5)-C(10) ring.